

TABLE 3
Experimental crystallization of Auckland Island olivine basalt at various pressures and temperatures

| Pressure (kb) | Temperature (°C) | Time (mins) | Sample capsule | Results |
|-------------------------------------|------------------|-------------|----------------|--|
| <i>A. Dry conditions</i> | | | | |
| 9.0 | 1280 | 30 | Pt | Olivine+glass. Very near liquidus |
| 9.0 | 1260 | 30 | Pt | Olivine+clinopyroxene+glass |
| 11.3 | 1320 | 30 | Pt | Above liquidus |
| 11.3 | 1310 | 30 | Pt | Clinopyroxene+rare orthopyroxene+minor olivine. Clinopyroxene with rare parallel growth of orthopyroxene |
| 13.5 | 1350 | 30 | Pt | Above liquidus |
| 13.5 | 1330 | 30 | Pt | Uncommon orthopyroxene and clinopyroxene+glass. Very near liquidus |
| 13.5 | 1330 | 60 | Graphite | Clinopyroxene+orthopyroxene+glass Clinopyroxene > orthopyroxene |
| 13.5 | 1325 | 60 | Pt | Orthopyroxene+glass |
| 13.5 | 1320 | 30 | Pt | Clinopyroxene+rare orthopyroxene+glass Cpx ≫ Opx |
| 13.5 | 1320 | 60 | Graphite | Clinopyroxene+orthopyroxene+glass Cpx > Opx |
| 13.5 | 1300 | 30 | Pt | Clinopyroxene+minor orthopyroxene+glass (~30% crystallization) |
| 18.0 | 1380 | 60 | Graphite | Above liquidus |
| 18.0 | 1370 | 60 | Graphite | Clinopyroxene+glass. Cpx may be quench |
| 18.0 | 1360 | 60 | Graphite | Clinopyroxene+possible rare orthopyroxene+glass |
| <i>B. "Wet" conditions</i> | | | | |
| 13.5 | 1200 | 30 | Pt | Olivine+glass. Very near liquidus |
| 13.5 | 1190 | 30 | Pt | Olivine+orthopyroxene+glass. Opx > ol |
| 13.5 | 1180 | 30 | Pt | Olivine+orthopyroxene+glass. Opx ≫ ol |
| 13.5 | 1160 | 30 | Pt | Olivine+orthopyroxene+clinopyroxene+glass. Opx > Cpx. Minor olivine. Possible amphibole |
| 13.5 | 1150 | 30 | Pt | Olivine+orthopyroxene+clinopyroxene+glass |
| 13.5 | 1130 | 30 | Pt | Olivine+orthopyroxene+amphibole+glass. Clinopyroxene not certain |
| 18.0 | 1260 | 30 | Pt | Orthopyroxene+glass. Very near liquidus |
| 18.0 | 1240 | 30 | Pt | Orthopyroxene+clinopyroxene+glass. Opx > Cpx |
| 18.0 | 1200 | 30 | Pt | Orthopyroxene+clinopyroxene+glass. Orthopyroxene and clinopyroxene intergrowths well developed Opx ≈ Cpx |
| <i>C. Controlled Water Contents</i> | | | | |
| 15.3 | 1200 | 20 | Pt | With 3% H ₂ O. Above liquidus |
| 14.4 | 1200 | 30 | Pt | With 2% H ₂ O. Above liquidus |
| 14.4 | 1170 | 30 | Pt | With 2% H ₂ O. Orthopyroxene+clinopyroxene+rare olivine. Opx common |
| 15.3 | 1200 | 20 | Pt | With 2% H ₂ O. Rare large orthopyroxene, no definite olivine. |
| 15.3 | 1170 | 30 | Pt | With 2% H ₂ O. Common orthopyroxene and possible rare olivine |

stricted field of orthopyroxene crystallization. The analytical data on the pyroxene compositions (table 4) demonstrate a very high degree of hypersthene solid solution in the clinopyroxene and somewhat lower Al₂O₃ contents in both pyroxenes than those observed in the previous olivine basalt. The clinopyroxenes have compositions suggestive of very magnesian pigeonites but comparison with the sub-calcic (9-10% CaO) clinopyroxenes previously obtained experimentally does not provide any evidence as yet for a compositional break

between augites, sub-calcic augites and compositions near to pigeonite. The coexistence of orthopyroxenes with slightly varying CaO content with this range of clinopyroxene compositions suggests that we are dealing with the "roof" of the two-pyroxene miscibility gap, the orthopyroxene side being "steep" (i.e. CaO content varies only slightly with temperature and with Mg/(Mg+Fe⁺⁺) ratio) while the clinopyroxene side in contrast is "shallow" and the hypersthene solid solution rapidly increases for small increases in temperature

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